

An inhibitor of Janus kinase 3: 4-(4-hydroxyphenylamino)-6,7-di- methoxyquinazolin-1-ium chloride methanol solvate

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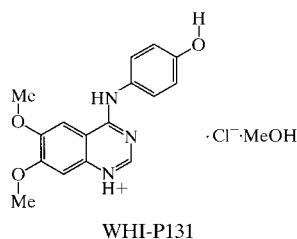
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The crystal structure of the title compound, $C_{16}H_{16}N_3O_3^{+}\cdot Cl^{-}\cdot CH_4O$ (WHI-P131, an inhibitor of Janus kinase 3), contains four hydrogen bonds. There are two hydrogen bonds within the asymmetric unit, *i.e.* interactions between WHI-P131 OH and Cl^{-} , and between methanol and Cl^{-} . There is a third interaction between WHI-P131 NH and Cl^{-} (related by a 2_1 screw) and a fourth between WHI-P131 NH and methanol (related by an *n*-glide). The hydrogen-bond pattern for these interactions can be described by the first-level hydrogen-bond graph-set notation $D_1^1(2)D_1^1(2)D_1^1(2)D_1^1(2)$. The second-level graph-set notation (for combinations of two hydrogen bonds) was determined to be $D_2^1(3)D_2^1(3)D_2^2(4)D_2^2(9)D_2^2(14)C_2^1(9)$.

Comment

The title compound, WHI-P131 (Fig. 1), inhibited the kinase activity of Janus kinase 3 (JAK3), with an IC_{50} of $9.1\ \mu M$ (Sudbeck *et al.*, 1999). Although WHI-P131 inhibited JAK3, it did not inhibit the Janus kinases JAK1 and JAK2, the ZAP/SYK family tyrosine kinase SYK, the TEC family tyrosine kinase BTK, the SRC family tyrosine kinase LYN or the receptor family tyrosine kinase IRK, even at concentrations as



high as $350\ \mu M$. The relatively high potency and selectivity of WHI-P131 for JAK3 makes it a promising candidate for new treatment strategies against acute lymphoblastic leukemia, the most common form of childhood cancer. In addition to its antileukemic properties, WHI-P131 also shows clinical

potential for the treatment of mast-cell-mediated immediate hypersensitivity reactions and allergic disorders (Malaviya *et al.*, 1999).

The crystal structure of WHI-P131 contains four different hydrogen bonds: $N1-H1\cdots O4$, $N4-H4\cdots Cl1$, $O3-H3\cdots Cl1$ and $O4-H26\cdots Cl1$ (Fig. 2 and Table 2). Each of these can be described in graph-set notation (Bernstein *et al.*, 1990, 1995; Etter, 1990, 1991; Etter *et al.*, 1990) as $D_1^1(2)$ (Fig. 2). The second-level motif combining the $N1-H1\cdots O4$ and $N4-H4\cdots Cl1$ hydrogen bonds is $D_2^2(9)$, that combining

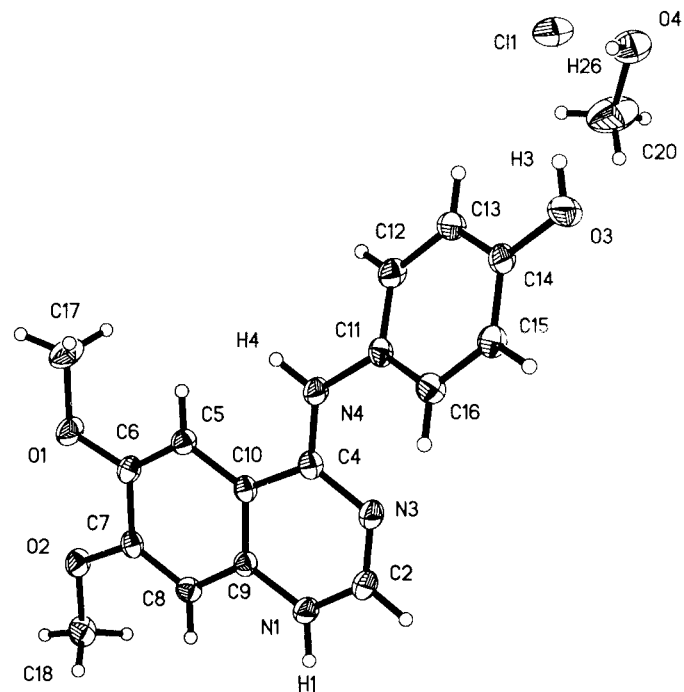


Figure 1
The X-ray crystal structure of WHI-P131 (30% probability displacement ellipsoids, $T = 297\ K$).

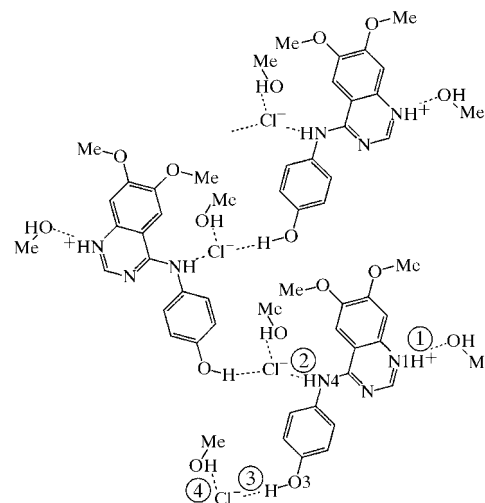


Figure 2
Hydrogen-bond patterns observed in WHI-P131. Four hydrogen bonds (labeled 1–4) are observed in the crystal structure. The complete first-level hydrogen-bond graph-set pattern is $D_1^1(2)D_1^1(2)D_1^1(2)D_1^1(2)$ and the second-level pattern is $D_2^1(3)D_2^1(3)D_2^2(4)D_2^2(9)D_2^2(14)C_2^1(9)$.

N4—H4···Cl1 and O3—H3···Cl1 is $C_2^1(9)$, and that combining N1—H1···O4 and O3—H3···Cl1 is $D_2^2(14)$. The combination of N1—H1···O4 and O4—H26···Cl1 can be described as a $D_2^2(4)$ pattern, N4—H4···Cl1 plus O4—H26···Cl1 forms a $D_2^2(3)$ pattern, and O3—H3···Cl1 plus O4—H26···Cl1 is $D_2^2(3)$. The complete first-level hydrogen-bond graph-set notation for WHI-P131 is $D_1^1(2)D_1^1(2)D_1^1(2)-D_1^1(2)$ and the second-level graph-set notation (for combinations of two hydrogen bonds) is $D_2^2(3)D_2^2(3)D_2^2(4)D_2^2(9)-D_2^2(14)C_2^2(9)$.

An alternative way to describe the two-dimensional hydrogen-bonded network in the crystal of WHI-P131 is $C_2^1(9)C_2^2(15)$, which combines the second-level motif for N4—H4···Cl1 and O3—H3···Cl1, $C_2^1(9)$, and the third-level motif for O4—H16···Cl1, O3—H3···Cl1 and N1—H1···O4, $C_2^2(15)$.

Experimental

Yellow needles of WHI-P131 were grown from methanol/dichloromethane by vapor diffusion at room temperature. The hydrochloride salt crystallized as a methanol solvate.

Crystal data

$C_{16}H_{16}N_3O_3^+ \cdot Cl^- \cdot CH_4O$
 $M_r = 365.81$
 Monoclinic, $P2_1/n$
 $a = 7.4128$ (7) Å
 $b = 10.7752$ (10) Å
 $c = 22.337$ (2) Å
 $\beta = 97.538$ (2)°
 $V = 1768.8$ (3) Å³
 $Z = 4$

$D_x = 1.374$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 2622 reflections
 $\theta = 2.64$ – 25.01 °
 $\mu = 0.243$ mm⁻¹
 $T = 297$ (2) K
 Needle, yellow
 $0.45 \times 0.15 \times 0.12$ mm

Data collection

CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: empirical
 (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.90$, $T_{\max} = 0.97$
 8982 measured reflections
 3117 independent reflections
 2087 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.050$
 $\theta_{\text{max}} = 25$ °
 $h = -8 \rightarrow 8$
 $k = -12 \rightarrow 12$
 $l = -26 \rightarrow 22$
 101 standard reflections
 intensity decay: -0.06%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.12$
 $S = 0.99$
 3117 reflections
 239 parameters

H atoms treated by a mixture of independent and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0672P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.003$
 $\Delta\rho_{\text{max}} = 0.33$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

N3—C2	1.303 (3)	O4—C20	1.404 (4)
N3—C4	1.356 (3)	N4—C4	1.330 (3)
O3—C14	1.375 (3)	N4—C11	1.433 (3)
C17—O1—C6—C5	7.6 (4)	C4—N4—C11—C16	-38.2 (4)
C18—O2—C7—C8	-5.4 (3)		

Table 2

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O4—H26···Cl1			3.024 (2)	
O3—H3···Cl1			3.064 (2)	
N1—H1···O4 ⁱ	0.93 (3)	1.81 (3)	2.742 (3)	179 (2)
N4—H4···Cl1 ⁱⁱ	0.88 (3)	2.39 (3)	3.206 (2)	155 (2)

Symmetry codes: (i) $x - \frac{1}{2}, \frac{1}{2} - y, \frac{1}{2} + z$; (ii) $\frac{1}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$.

Table 3

First- and second-level graph-set motifs for hydrogen bonds in WHI-P131.

	O4—H26···Cl1	O3—H3···Cl1	N1—H1···O4	N4—H4···Cl1
O4—H26···Cl1	$D_1^1(2)$	$D_2^2(3)$	$D_2^2(4)$	$D_2^2(3)$
O3—H3···Cl1		$D_1^1(2)$	$D_2^2(14)$	$C_2^2(9)$
N1—H1···O4			$D_1^1(2)$	$D_2^2(9)$
N4—H4···Cl1				$D_1^1(2)$

H atoms were placed at calculated positions, except for H1 and H4, which were located in the electron-density difference map and were refined isotropically. The hydroxyl H3 and H26 atoms were not observed in the electron-density map but were included at calculated positions based on an assessment of the best hydrogen-bond interactions to nearby N, O or Cl atoms.

Data collection: SMART (Bruker, 1998); cell refinement: SMART; data reduction: SHELXTL (Bruker, 1998); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1527). Services for accessing these data are described at the back of the journal.

References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
 Bernstein, J., Etter, M. C. & MacDonald, J. C. (1990). *J. Chem. Soc. Perkin Trans. 2*, pp. 695–698.
 Bruker (1998). SAINT, SMART and SHELXTL/NT (Version 5.1) Software Reference Manuals. Bruker AXS Inc., Madison, Wisconsin, USA.
 Etter, M. C. (1990). *Acc. Chem. Res.* **23**, 120–126.
 Etter, M. C. (1991). *J. Phys. Chem.* **95**, 4601–4610.
 Etter, M. C., MacDonald, J. C. & Bernstein, J. (1990). *Acta Cryst.* **B46**, 256–262.
 Malaviya, R., Zhu, D., Dibirdik, I. & Uckun, F. M. (1999). *J. Biol. Chem.* **274**, 27028–27038.
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
 Sudbeck, E. A., Liu, X.-P., Narla, R. K., Mahajan, S., Ghosh, S., Mao, C. & Uckun, F. M. (1999). *Clin. Cancer Res.* **5**, 1569–1582.